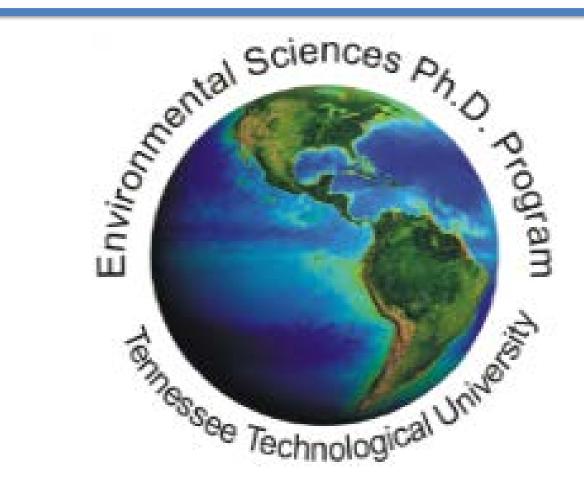


Degradation of Organic Contaminates from wastewater by Photocatalytic Method via TiO2 Thin film-Preliminary Investigation (With Simultaneous Production Hydrogen)



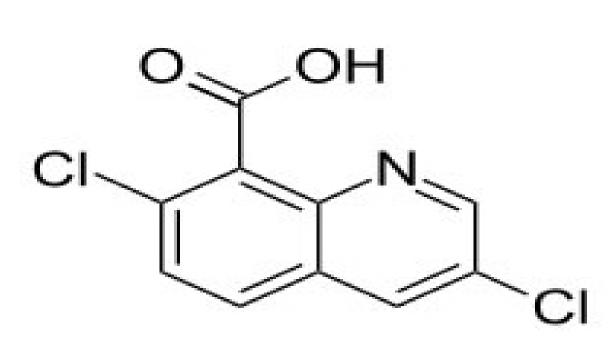
Sunil Rawal*, Njideka H. Okoye, S.Mahajan and Pedro Arce
College of Interdisciplinary studies
Department of Chemistry (*) and Department of Chemical Engineering
Tennessee Technological University, Tennessee

Objective and advantage

- 1.Elimination of filtration steps in water treatment with TiO₂.
- 2. Simultaneous production of H_2 from the same procedure.
- 3. Scaling of process to meet demands.

Model Treatment

Quinclorac (3,7-Dichloro-8-quinoline carboxylic acid) selective auxine herbicides applied for paddy field mainly used to control bardyard grass *Echinochloa Crus-galli*



Impact of this compound

- 1. Ground Water Contamination
- 2. Phyto toxicity to succeeding crops (Tomato, potato, egg plant etc.)
- 3. Toxicity to Fish and aquatic animals

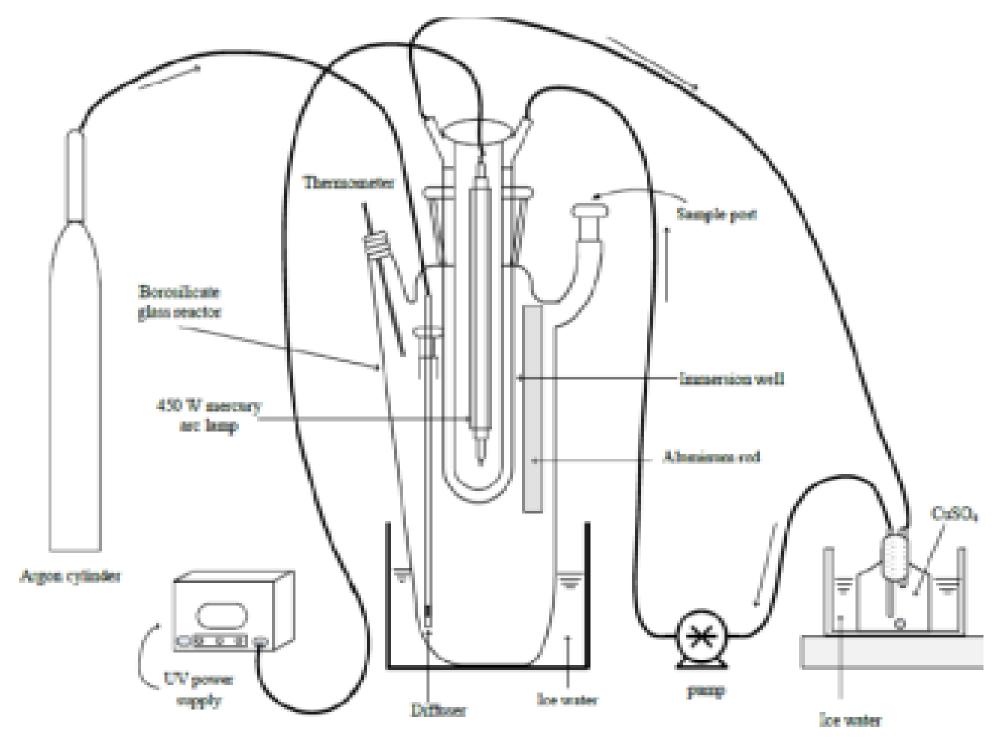


Fig. 1 UV Photocatalytic reactor set up

Materials and Methods

- 1.The advanced oxidation processes (AOPs) helps for conversion of many organic compounds (herbicides) by degradation
- 2.Degradation of herbicides using Mercury UV vapor lamp 450 W light based on Photocatalytic reactor using TiO₂ in coated film.
- 3.TiO₂ thin coated film deposited on film is characterized by X-ray diffraction (XRD), SEM and TEM.
- 2.Coated film tested using UV-VIS spectrophotometer.
- 4.The degradation of the herbicides would be assessed by liquid chromatography and mass spectrometry (LC- MS) to identify the compound concentration and intermediate degradation products.
- 5.All these reacts with organic contaminants that lead to effective decontamination.



Fig. 2 UV-Lamp
For Photo-cataly
Tic reactor

Increase in the percentage ethanol concentration causes increases in H_2 production rate with the highest rate of 21.2µmol H_2 / min at 120 min with 6 Pt-TiO₂ coated slides and 30% ethanol solution. Pt-TiO₂ surface area (from 4 slides to 6 slides) also increase H_2 production.

Results and Discussion

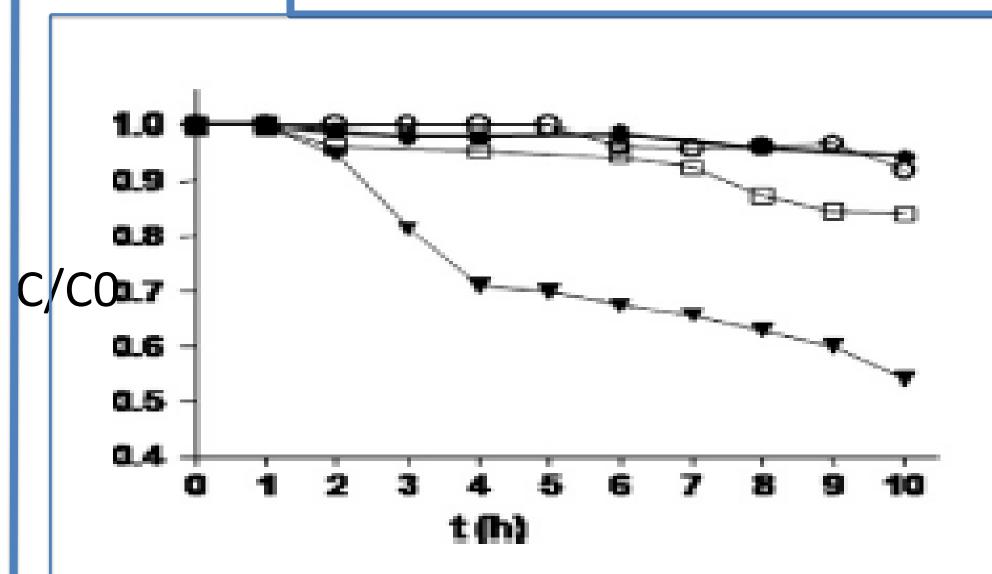


Fig.3 Deg. Of QNC in paddy field water under 250 and 700 Wm⁻²

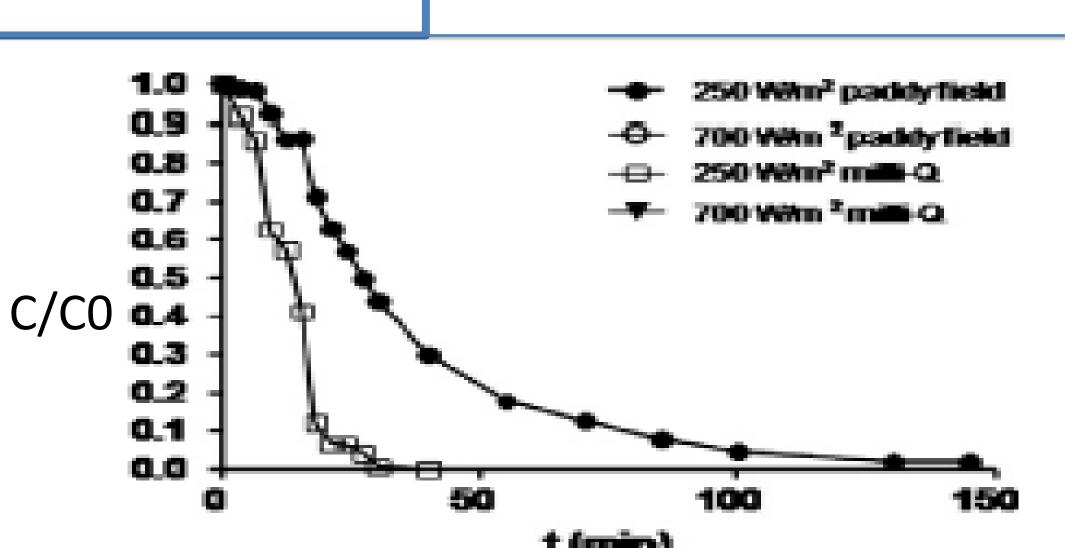
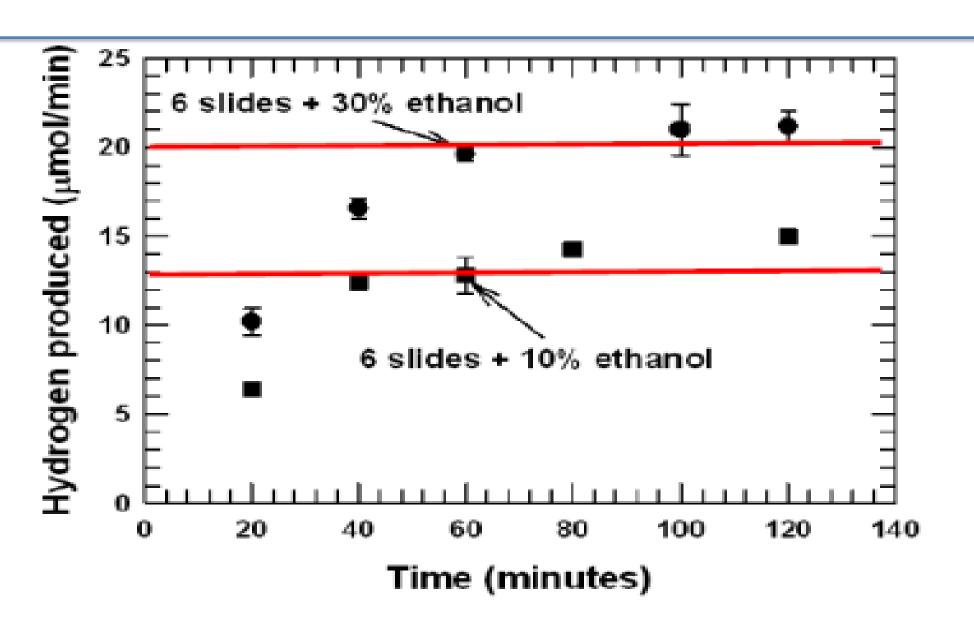


Fig.4 Deg. of QNC in pure and paddy water at 250Wm⁻²

Note: Fig. 3 and 4 from Reference No. 1



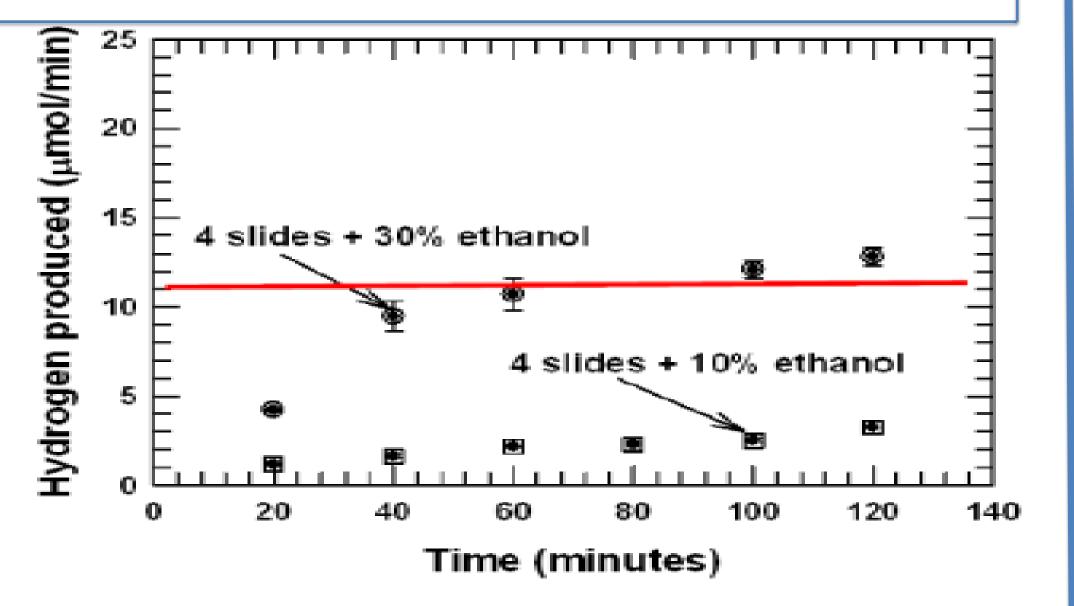
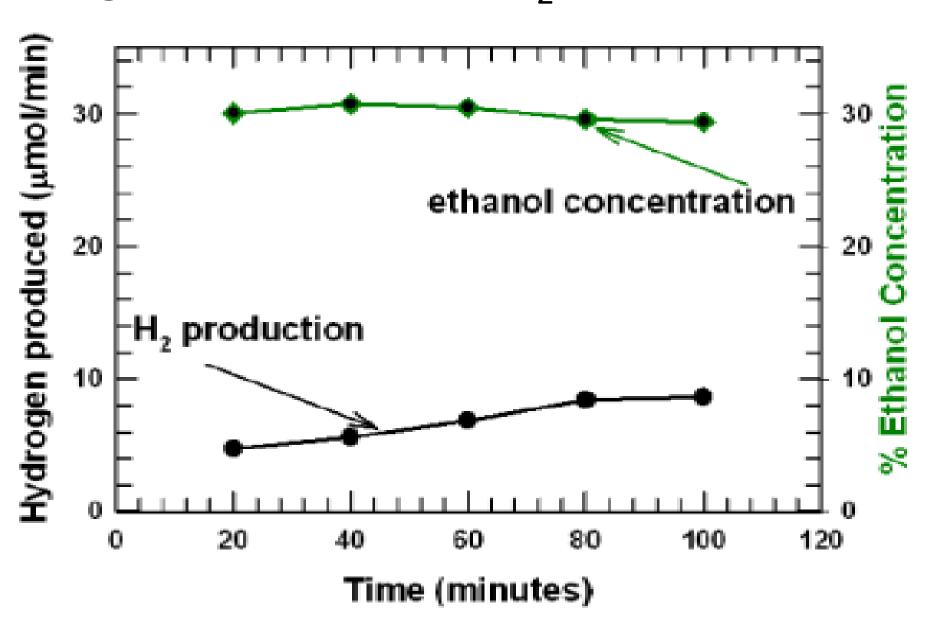
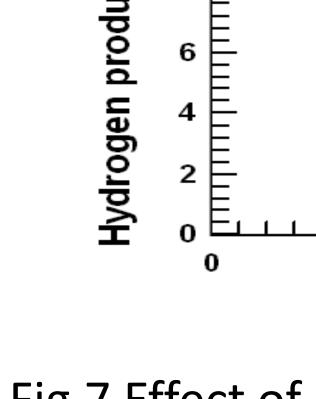


Fig .5 Effect of Pt- TiO₂ surface area and solution ethanol concentration on H₂ production





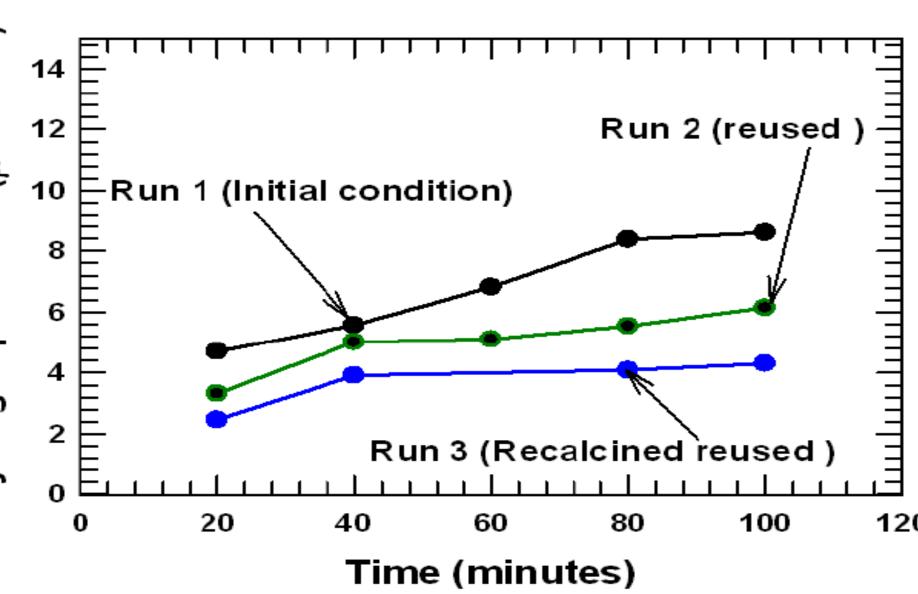


Fig.6 % Ethanol concentration With H₂ production rate

Fig.7 Effect of using different slide condition

References

- 1. Pareja L., Parada P.A., Aguera A., Cesio V., Hinzen H., Fernandez-Alba R.A., Photolytic and photocatalytic degradation of quinclorac in ultrapure and paddy field water: Identification of transformation products and pathways, *Chemosphere*, 2012(87)838-844
- 2.Okoye, N. H., Master Science Thesis, Tennessee Technological University, 2011